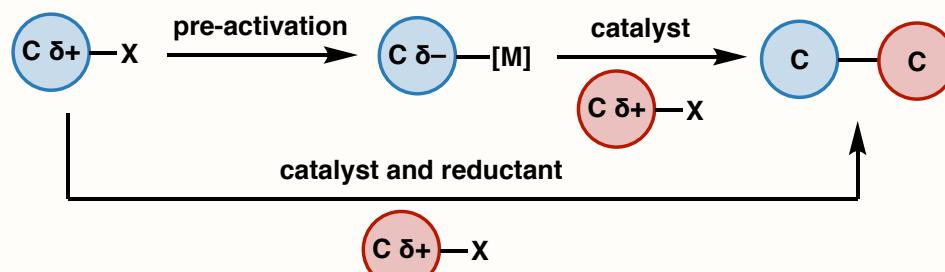


Cross electrophile coupling



XEC strategies

- xs of one reagent (not cross selective)
- Electronic differentiation/matching
- Steric differentiation/matching
- Distinct reactivity/mechanistic pathways

Cross electrophile coupling (XEC) advantages

- Cuts down one step (making nucleophile)
- Safety (avoids pyrophoric e.g. RMgX, or toxic e.g. RSnR₃)
- Easier to handle
- Increased stability/easier storage
- More commercial availability
- Better FG compatibility

Commercially available (2011)	
R-B(OH) ₂	~ 5500
R-I	~ 80000
R-Br	~ 700000

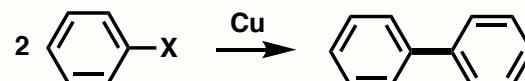
JACS 2012, 6146

XEC Disadvantages

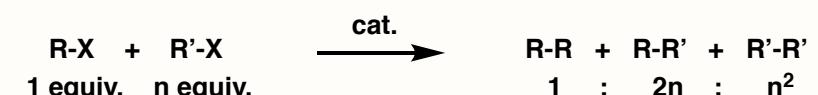
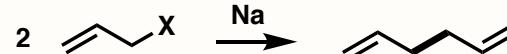
- Mostly heterogeneous (elemental metals and salts)
- Sometimes sensitive to stir rates and purity and mesh size of metal
- Inherently less selective

Electrophile coupling known for a long time: homodimerization

Ullman coupling



Wurtz coupling



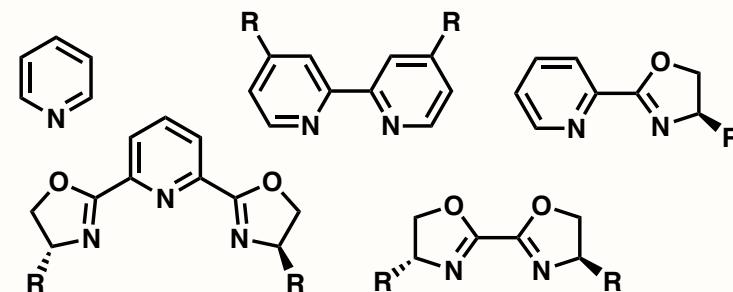
n	R-R' yield ^a	Pdt:dimers ^a
1	50%	1:1
2	80%	1:1.25
3	86%	1:1.67

^aAssuming complete rxn reversibility

JACS 2012, 6146

Stars of the show:

Ligands



Reductants

- Zinc
- Manganese
- Magnesium

Catalysts

Nickel (NiX₂)

Definition of XEC (for today)

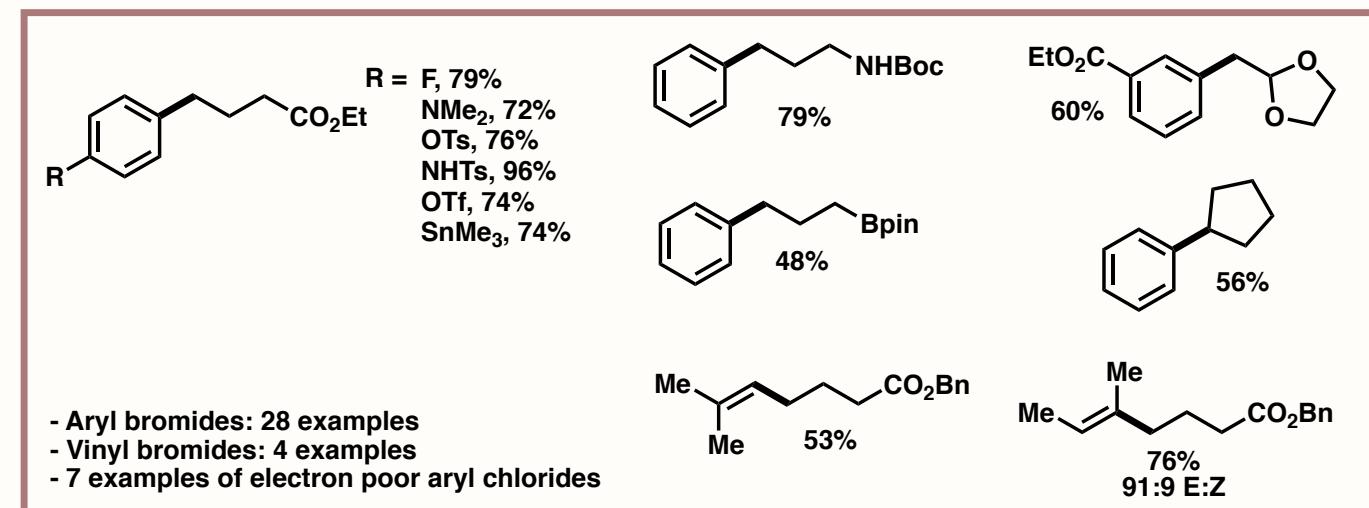
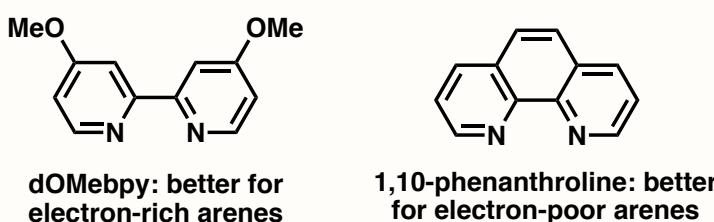
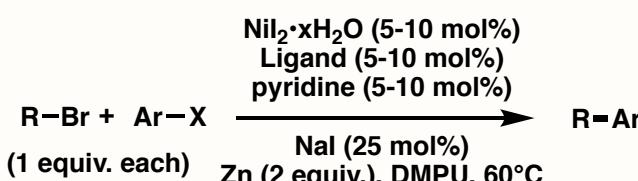
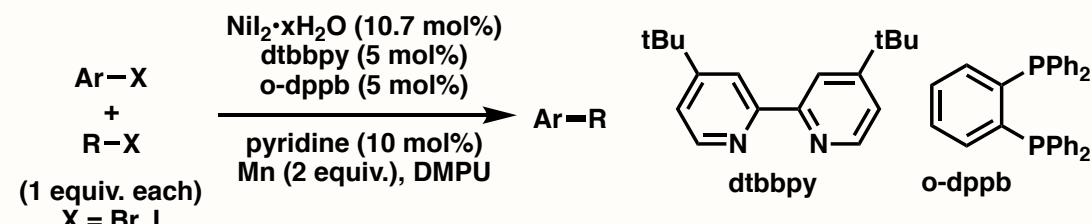
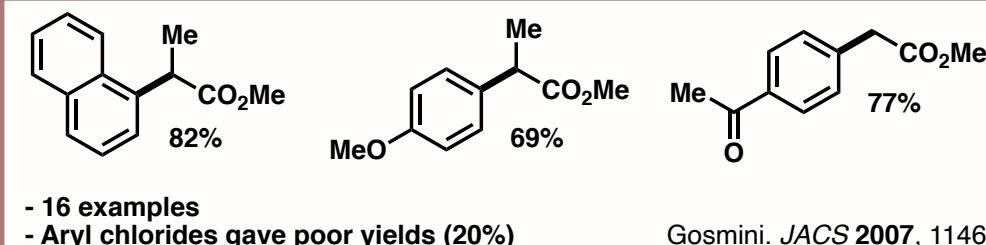
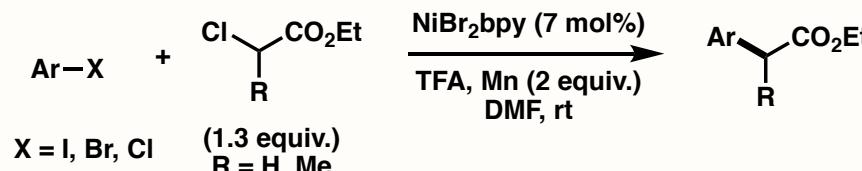
- No in situ formation of nucleophile
- Reductant acts on catalyst
- No addition into polar π systems (e.g. NHK)
- Will not be discussing electrochemical methods

Outline

1. Aryl-alkyl
2. Allylation
3. Vinylation
4. Alkyl-alkyl
5. Aryl-aryl
6. Acylation
7. Epoxide/aziridine opening

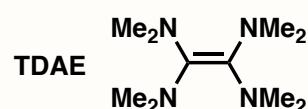
Cross electrophile coupling

Aryl-alkyl

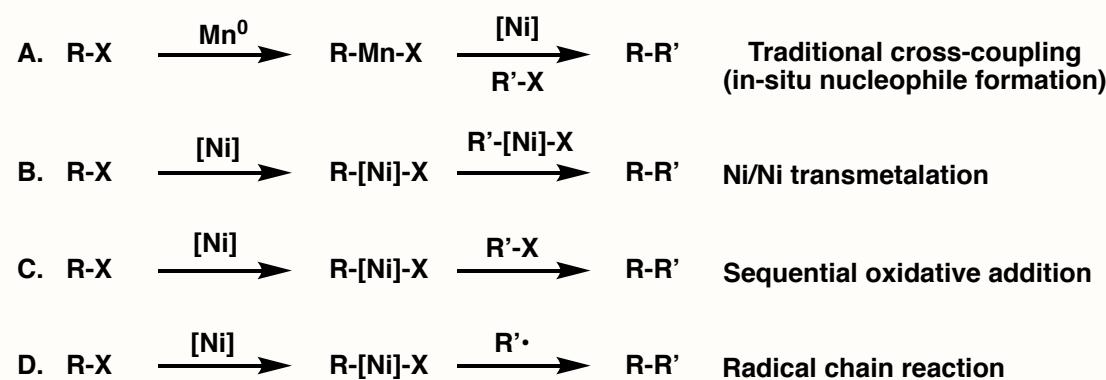


Mechanistic insights

- RZnX and ArZnX slow to form under rxn conditions
- TDAE instead of Zn gives 54% yield for a representative reaction

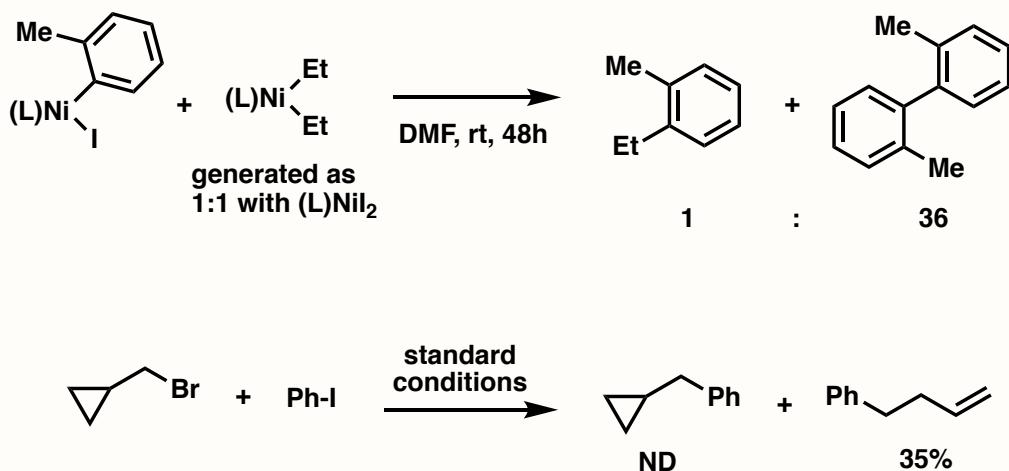


Cross electrophile coupling

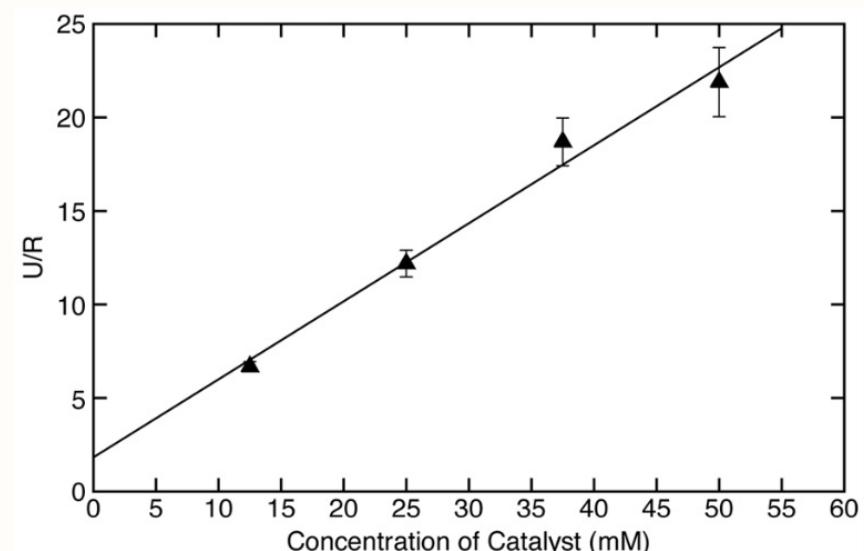
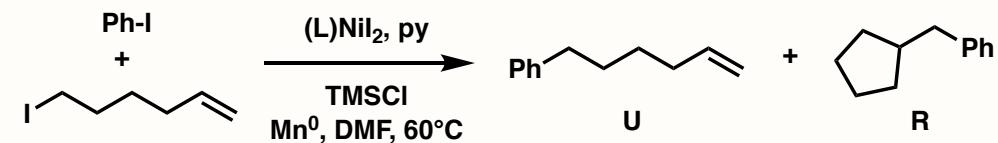
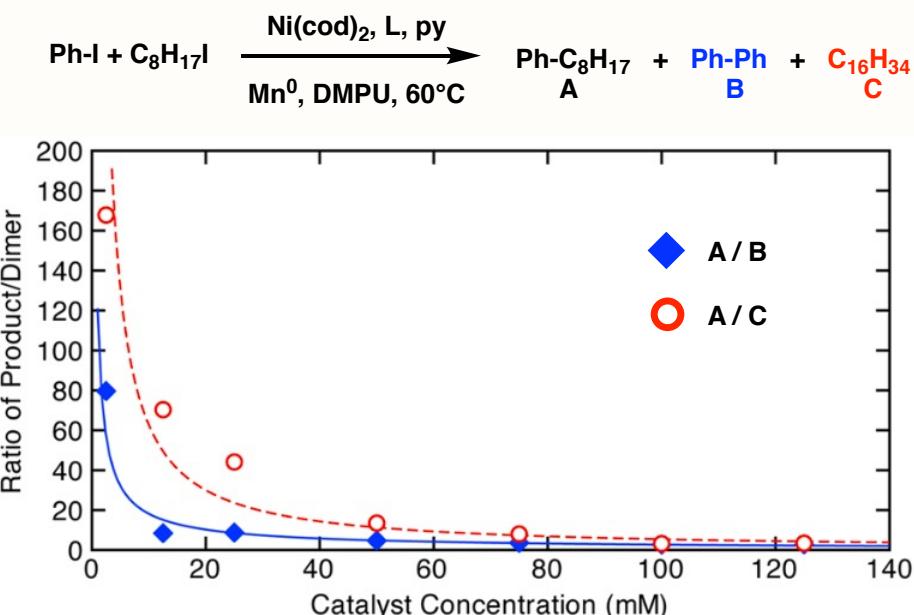


Quenching exp. and stoichiometric organonickel exp. supports initial ox. addn. of aryl halide, unfeasible for alkyl halide

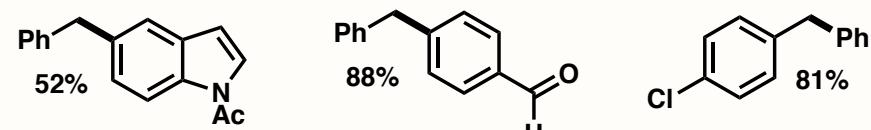
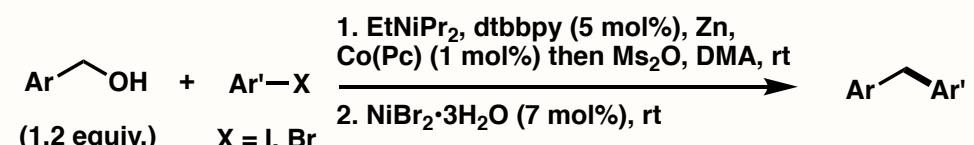
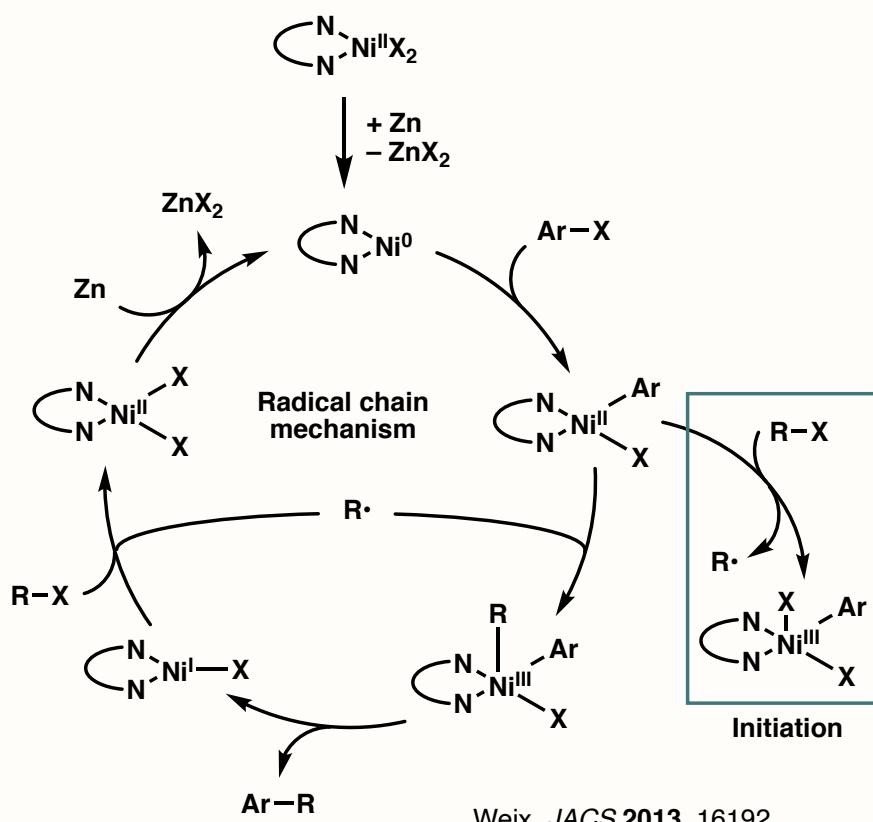
But Ni cat. biaryl formation has transmet. mechanism and is 2nd order in Ni
- J. Organomet. Chem. 1992, 223



Weix. JACS 2013, 16192



Cross electrophile coupling

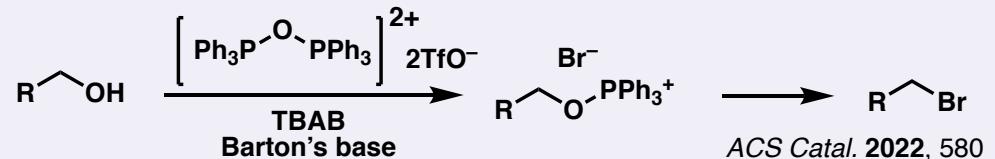
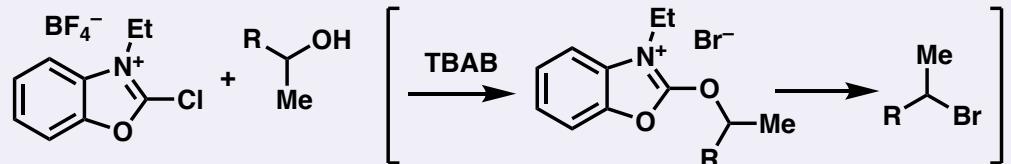
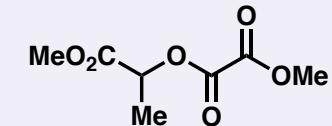


Weix. Chem. Sci. 2015, 1115

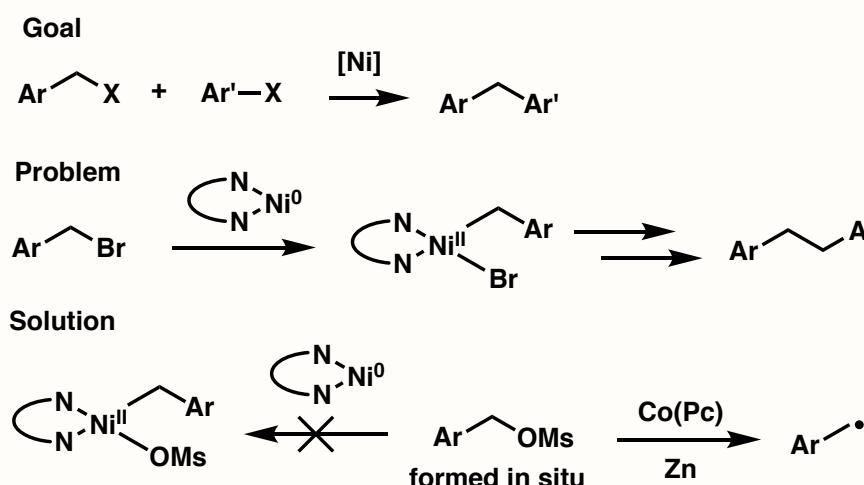
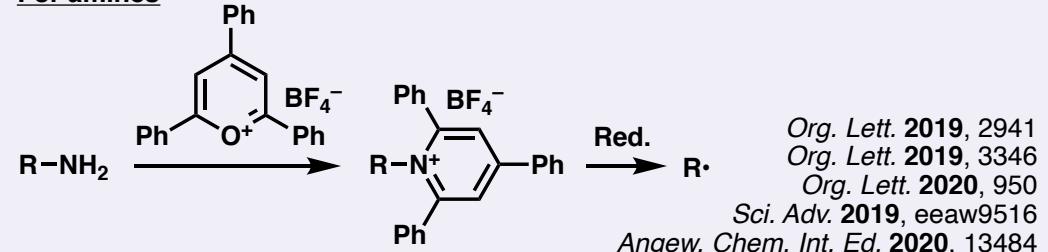
Other similar alcohol coupling strategies



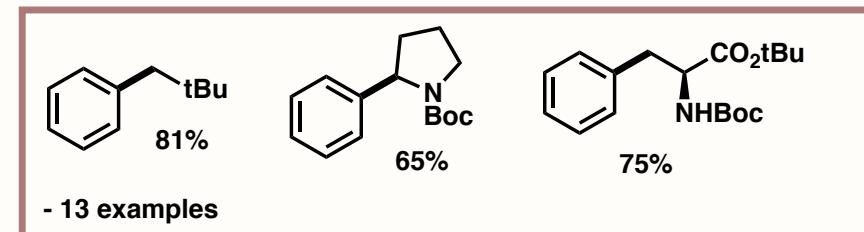
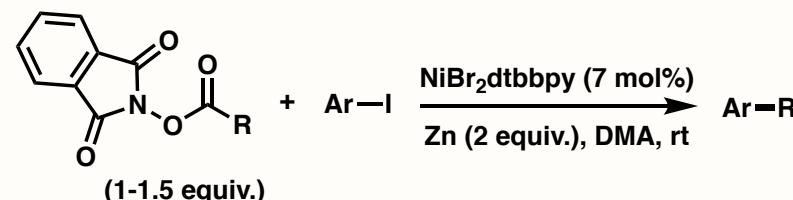
Org. Lett. 2018, 7846



For amines

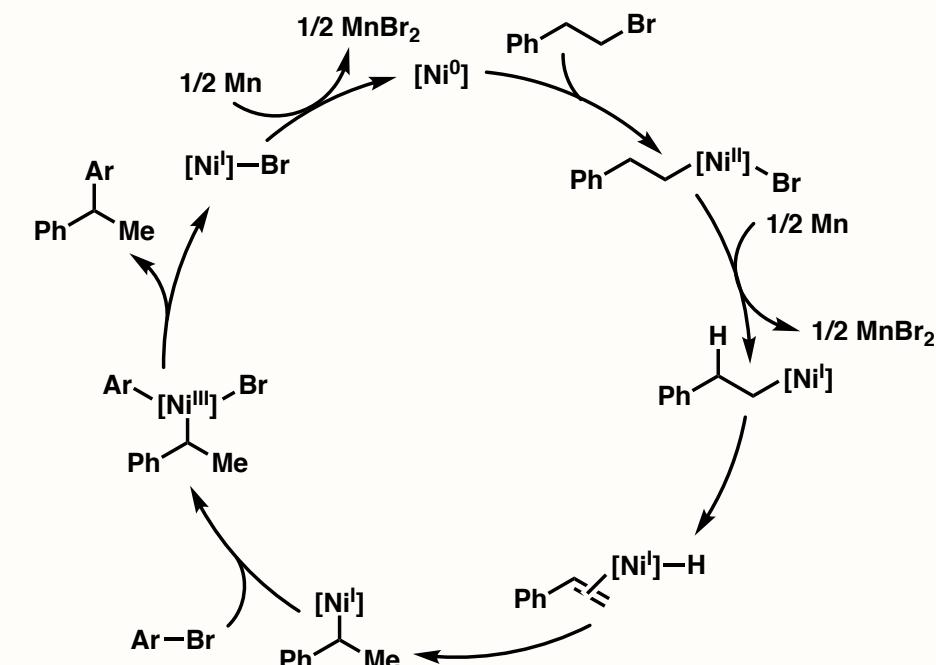
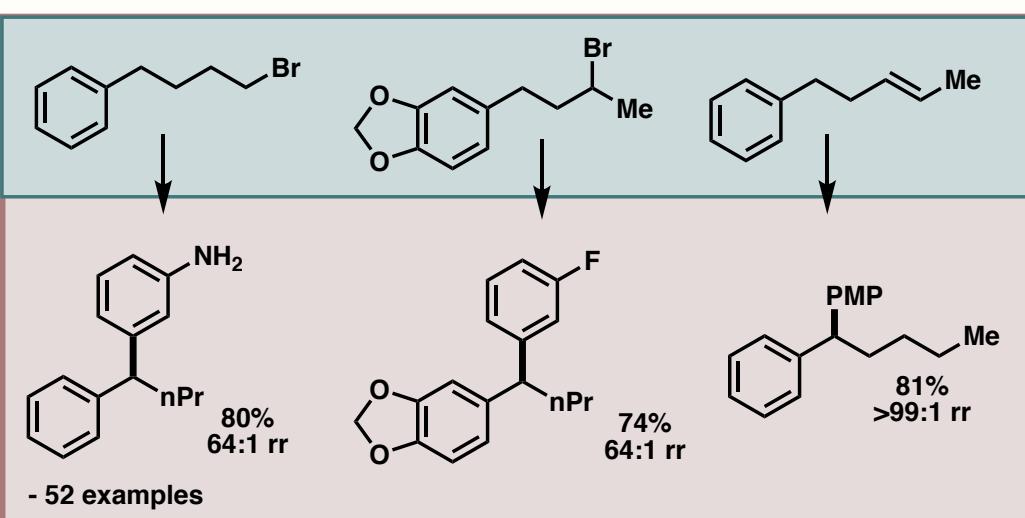
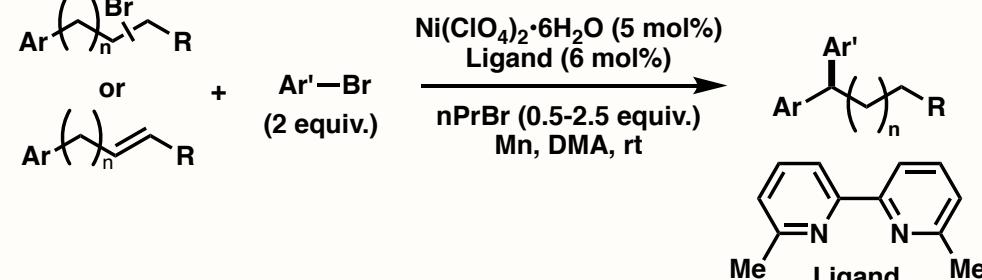
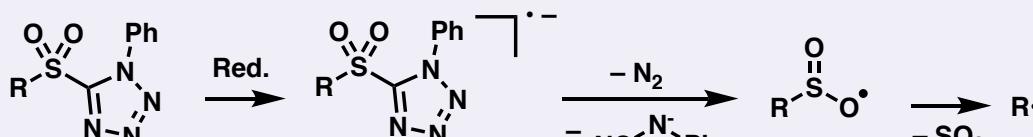


Cross electrophile coupling



Weix. JACS 2016, 5016

Similar strategy

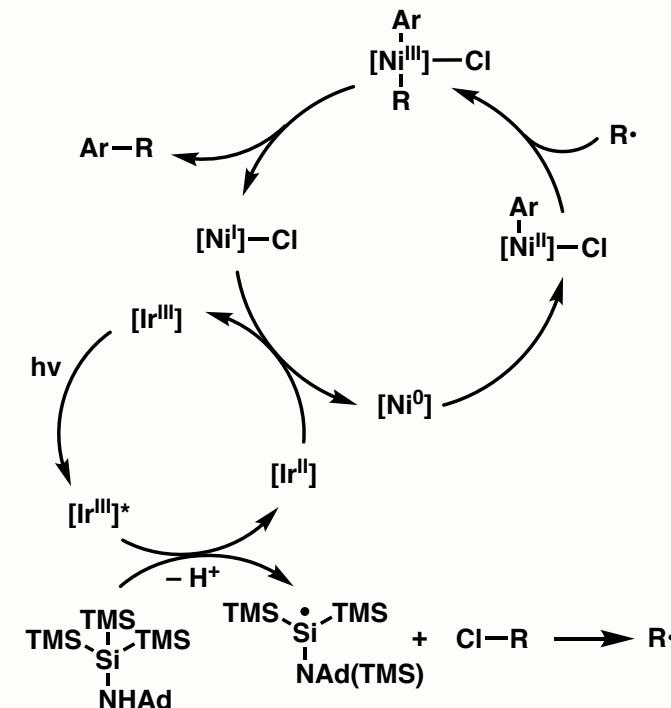
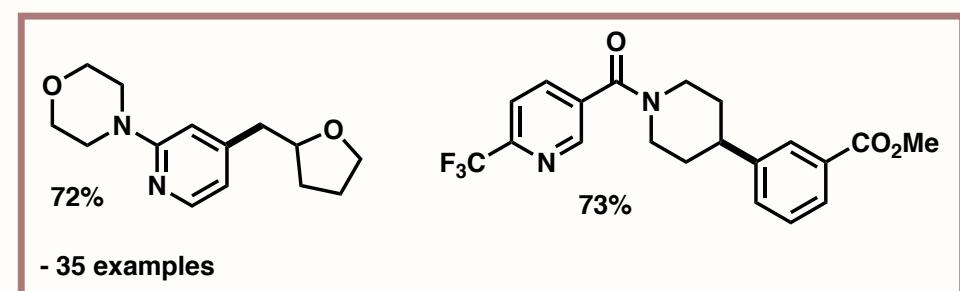
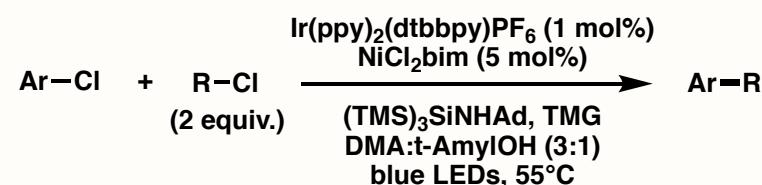
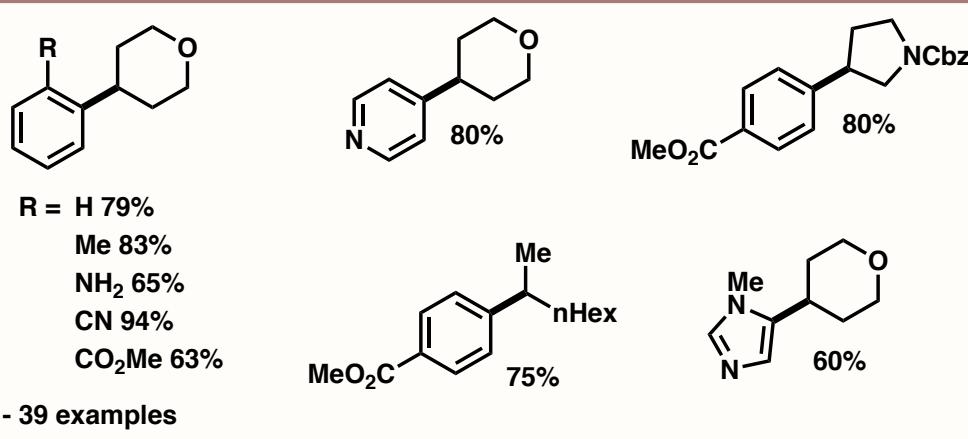
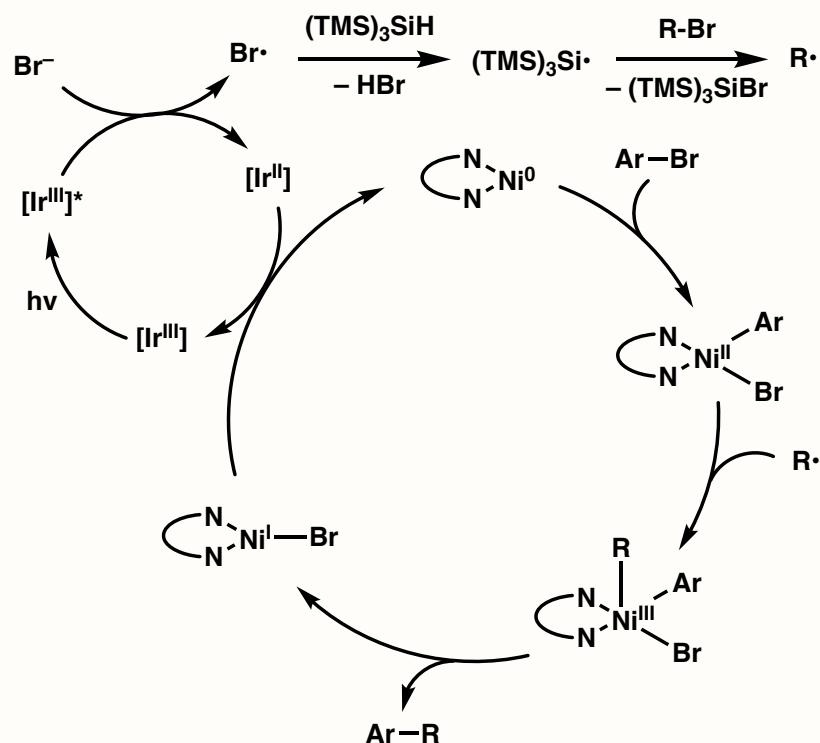
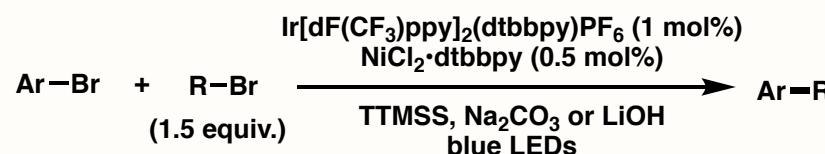


Zhu. JACS 2017, 13929

Similar chain walking XECs:

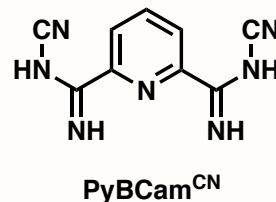
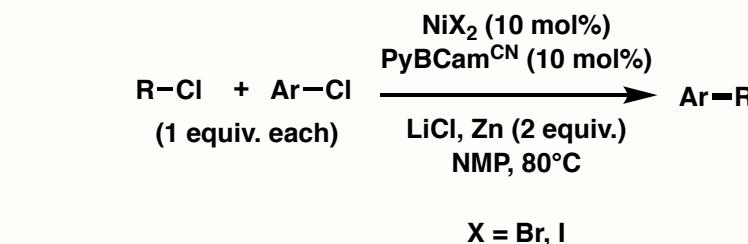
JACS 2017, 1061 (initial report uses ArI and generates NiH with CsF and PMHS)
ACS Catal. 2018, 310 (proposes initial OA to ArBr)

Cross electrophile coupling



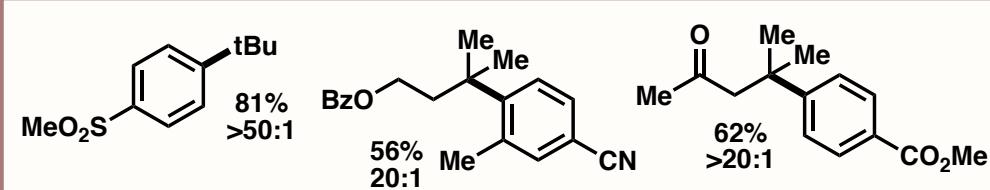
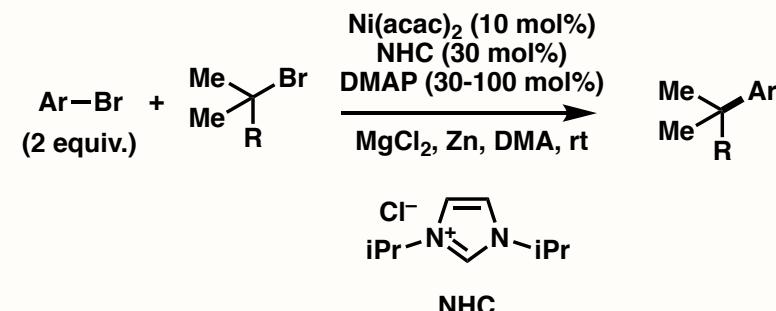
MacMillan. JACS 2020, 11691

Cross electrophile coupling



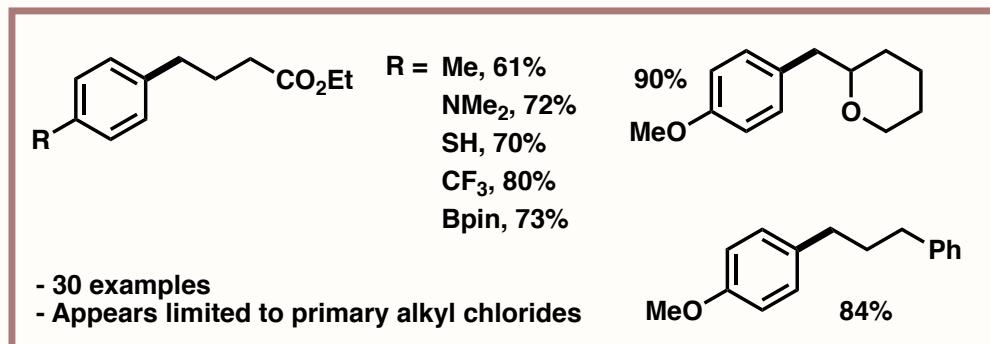
Crucial for reacting with both substrates at equal rate, suppressing both alkyl and aryl dimerization

Commercially available (2020)	
R-I	< 100000
R-Br	~ 841000
R-Cl	~ 1.8M

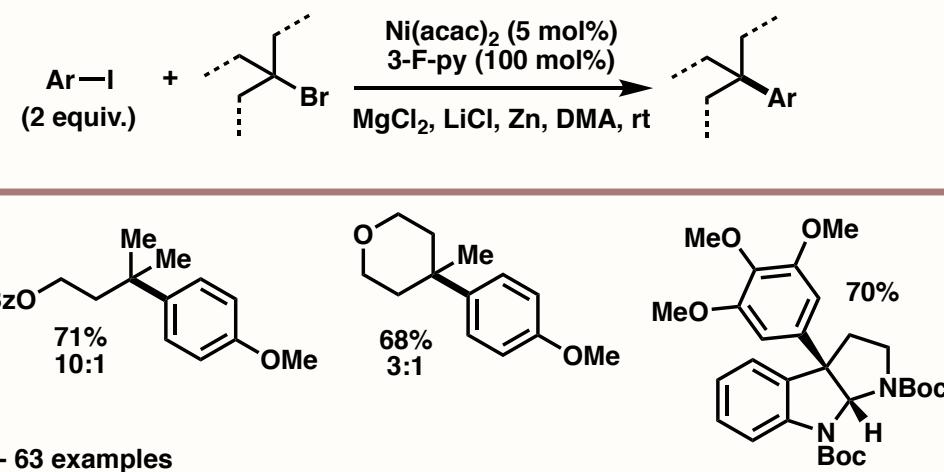


- 33 examples
- Only suitable for electron-poor arenes
- All yields within 12% of standard conditions if NHC omitted!
- Slight decrease in pdt/isomer without NHC

Gong. JACS 2015, 11562

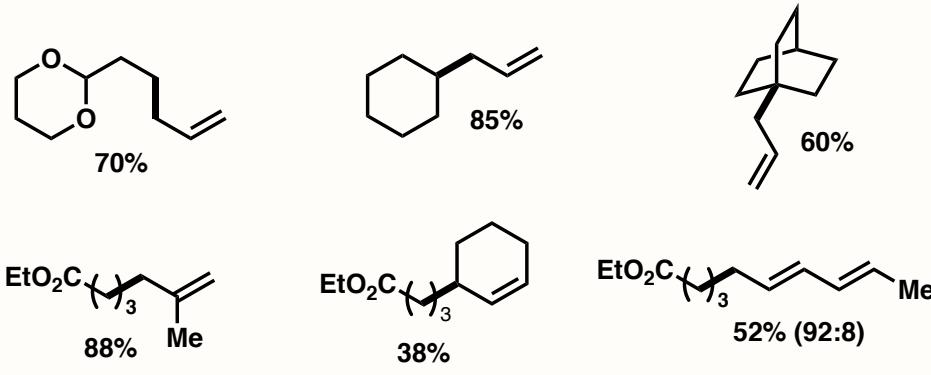
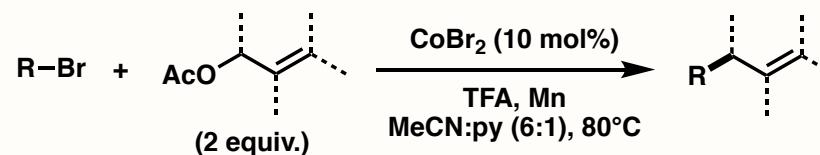
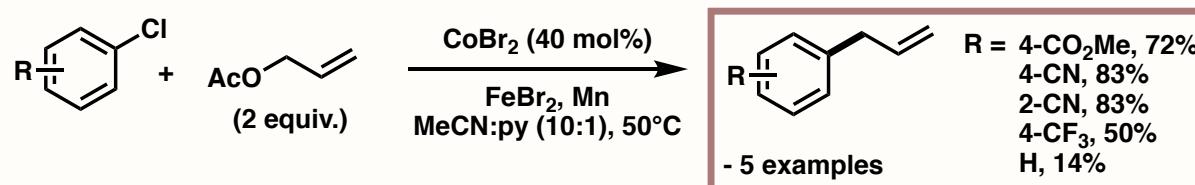


Weix. JACS 2020, 9902

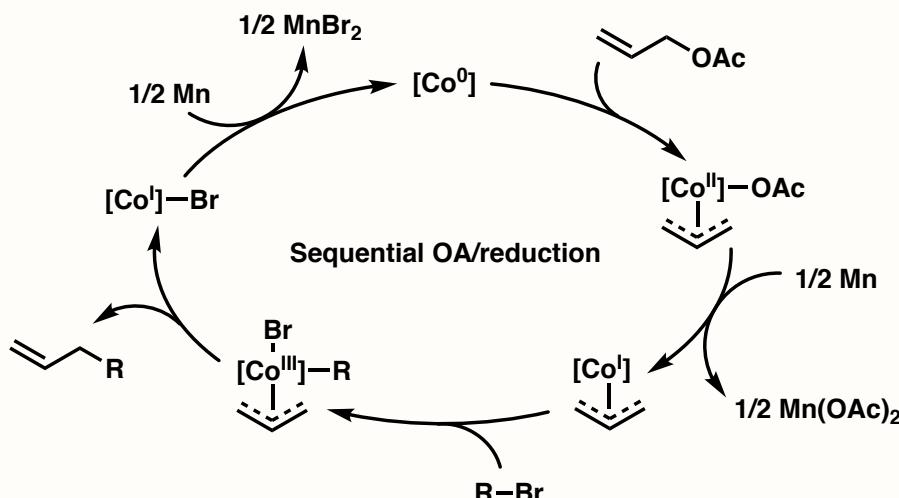


Gong. JACS 2018, 14490

Cross electrophile coupling



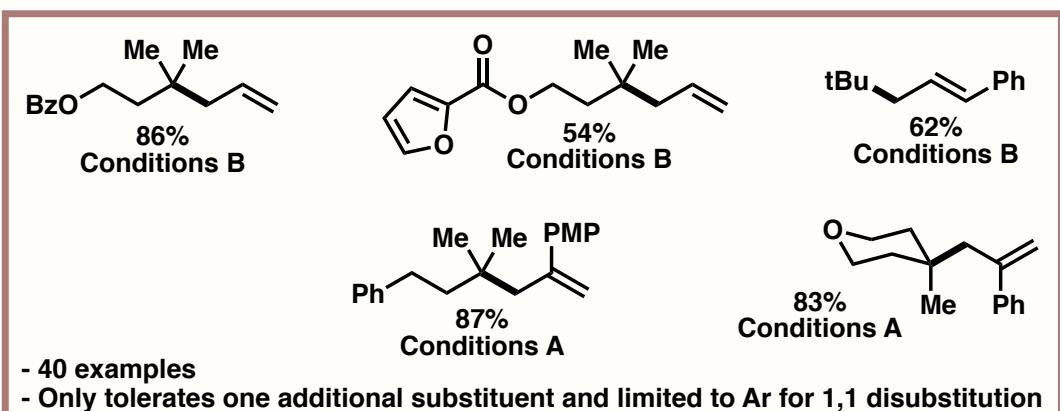
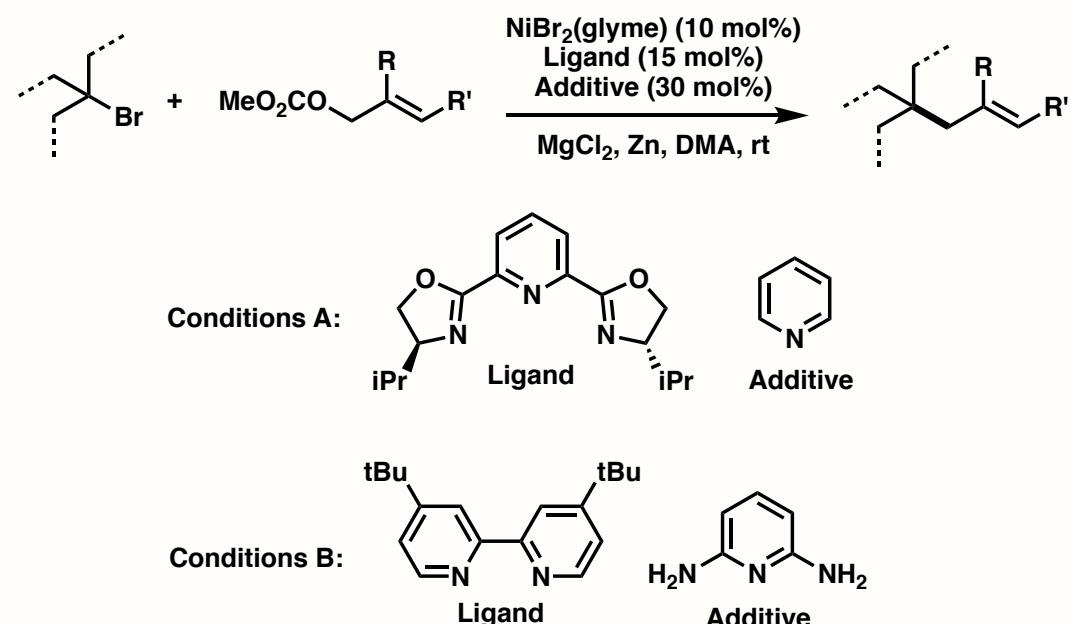
Gosmini. *Angew. Chem. Int. Ed.* 2011, 10402



Allylation

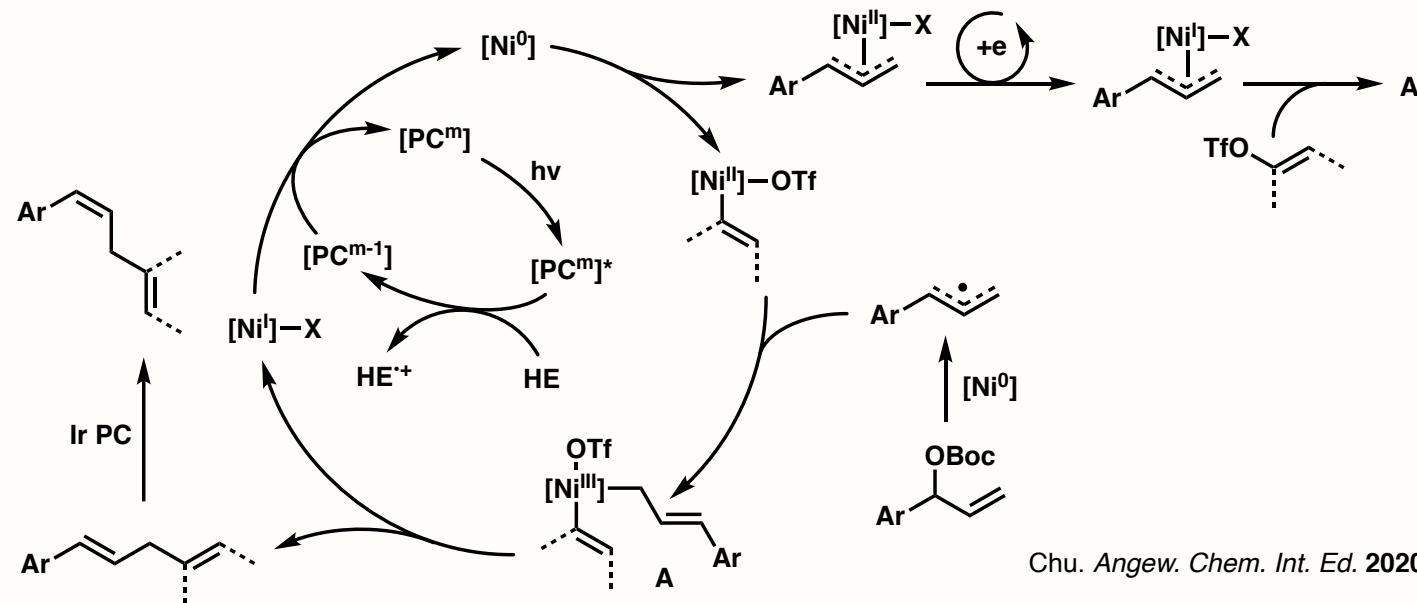
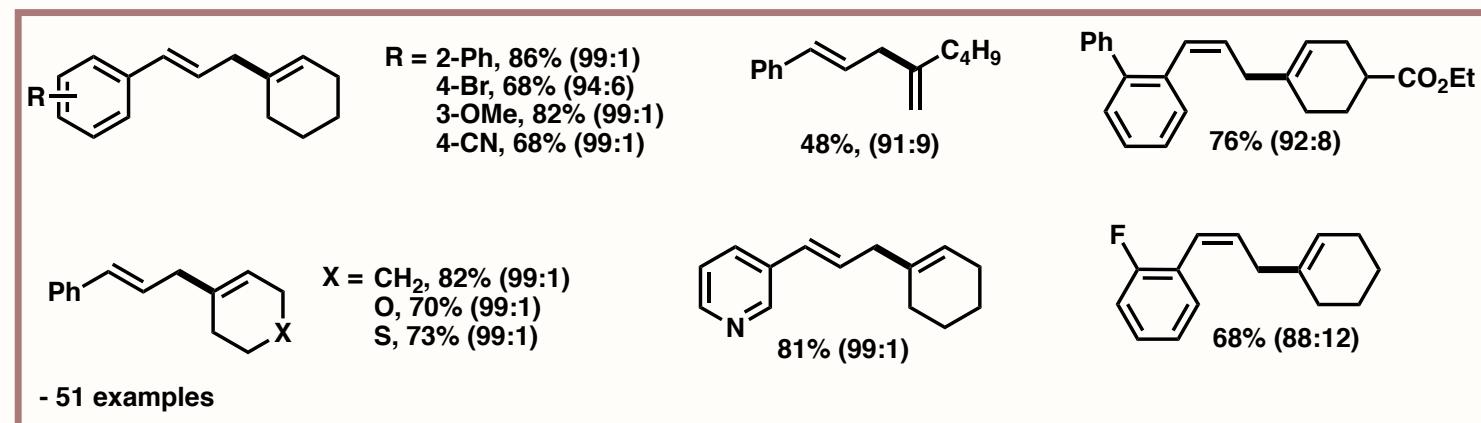
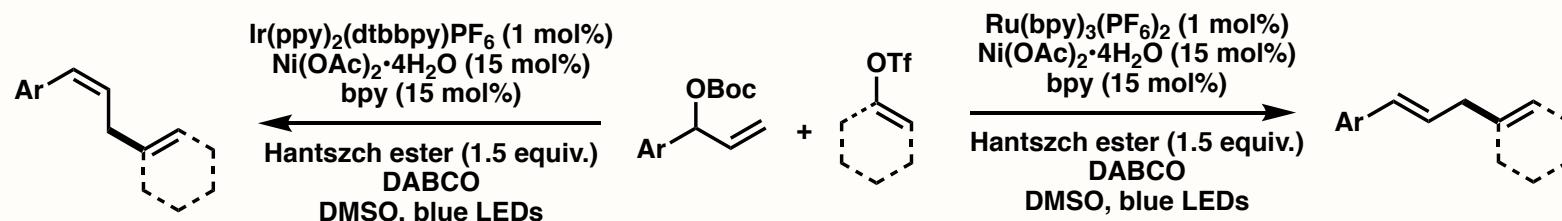
- Similar allylation with Ar-Br and Zn presumed to proceed through ArZnBr
- This allylation presumed a true XEC (no intermediate ArMnCl)
- Mechanism not speculated

Gosmini. *Org. Lett.* 2003, 1043



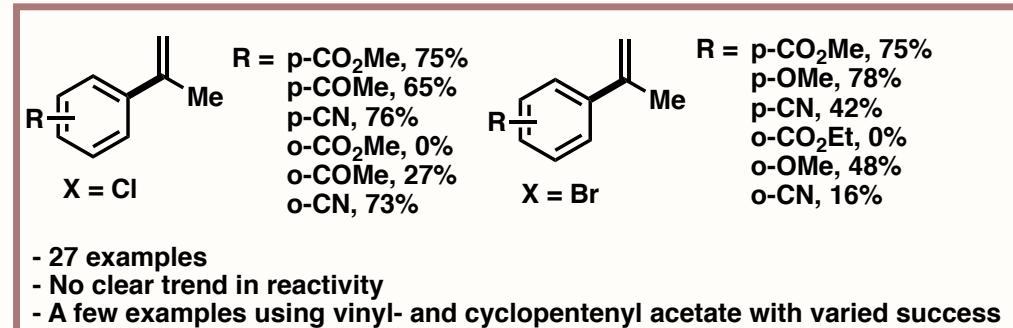
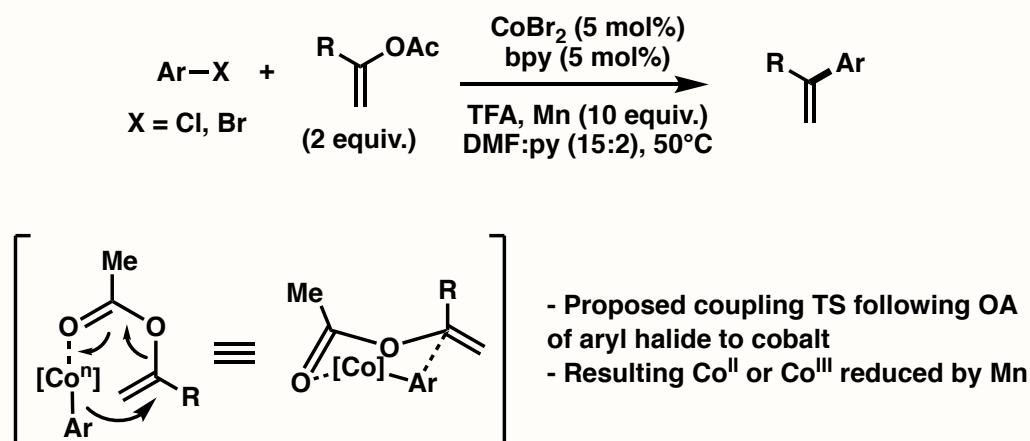
Gong. *Angew. Chem. Int. Ed.* 2017, 13103

Cross electrophile coupling

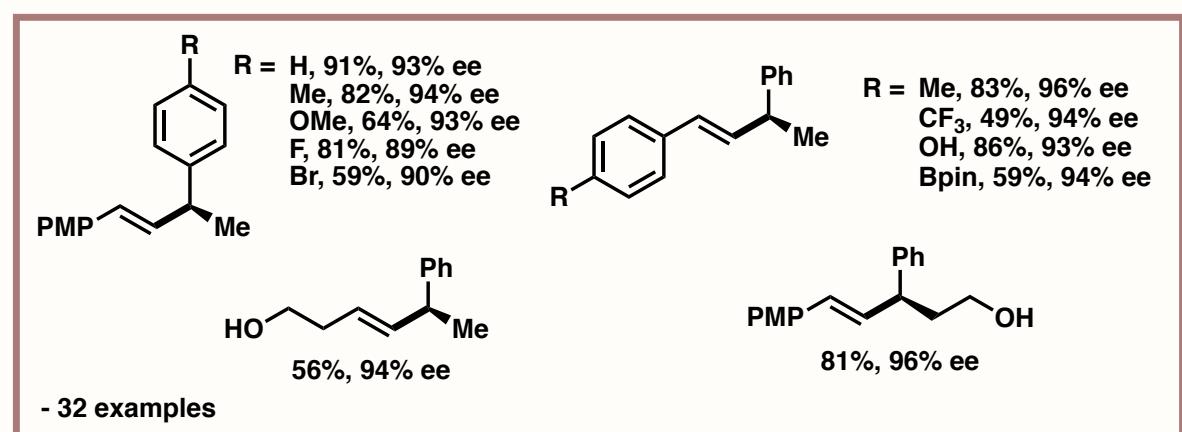
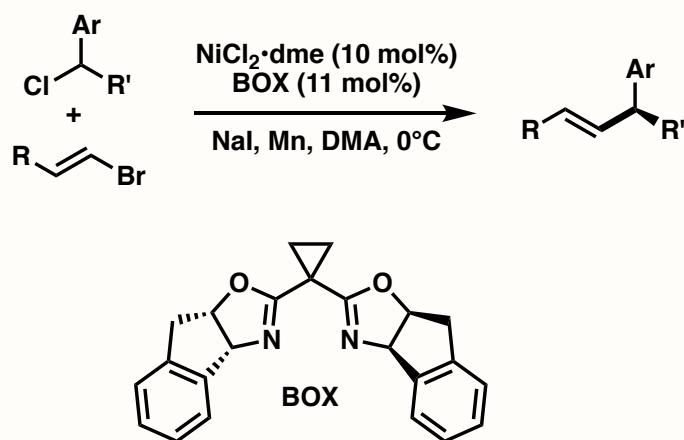


Cross electrophile coupling

Vinylation

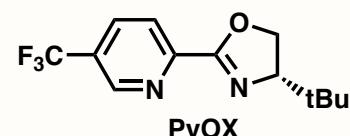
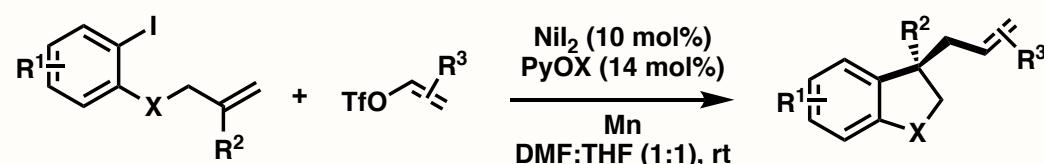


Gosmini. Eur. JOC 2005, 989

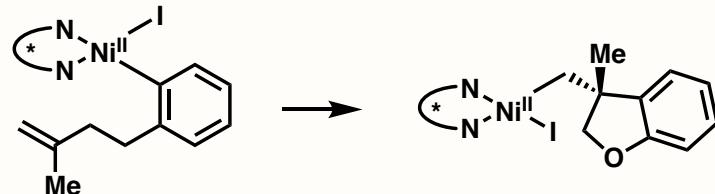
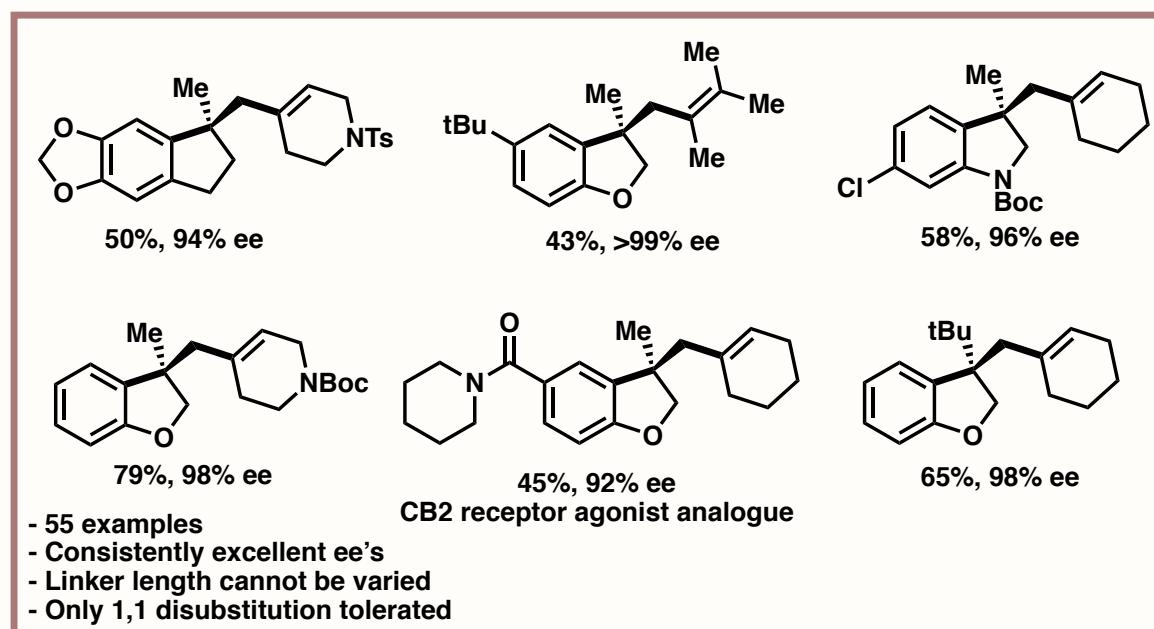
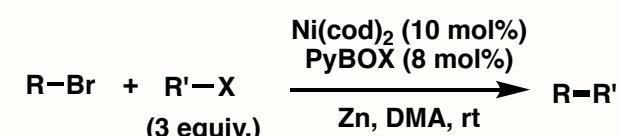


Reisman. JACS 2014, 14365

Cross electrophile coupling

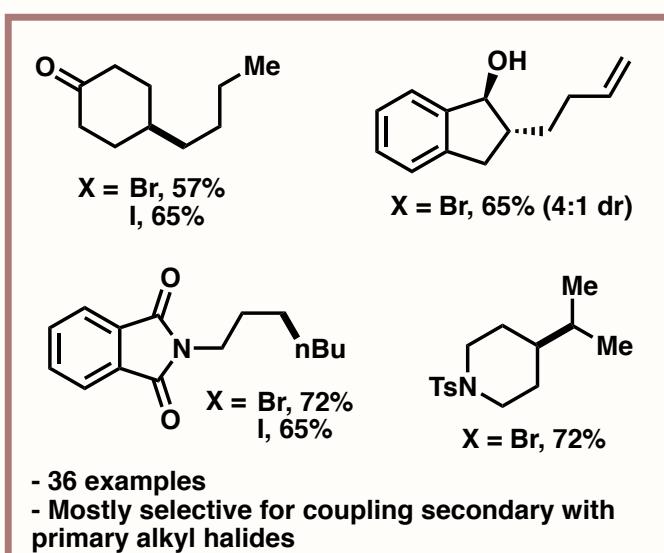
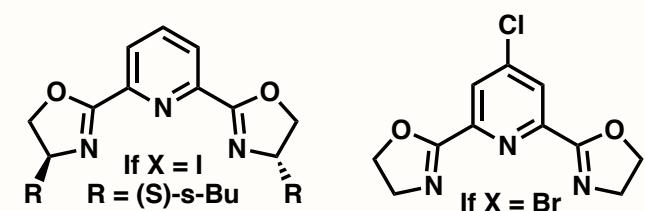


Alkyl-alkyl



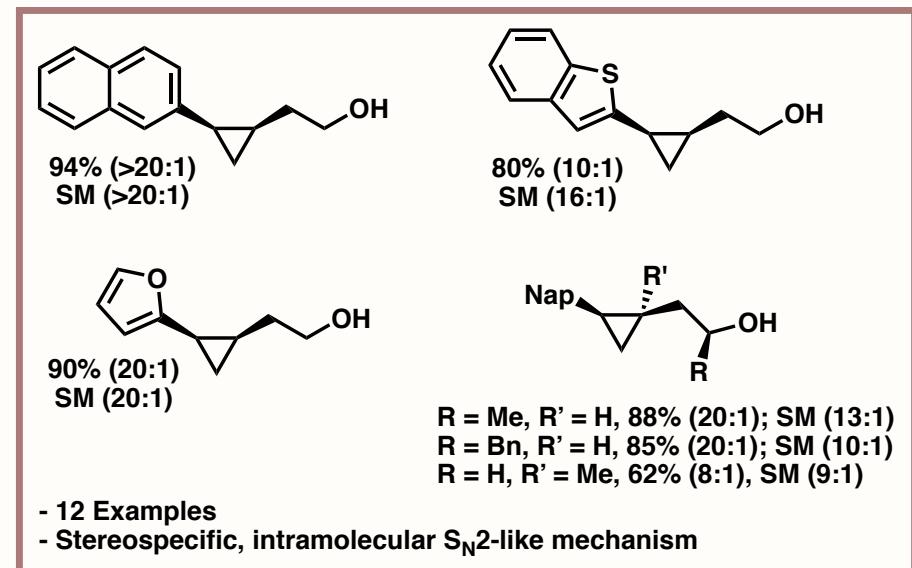
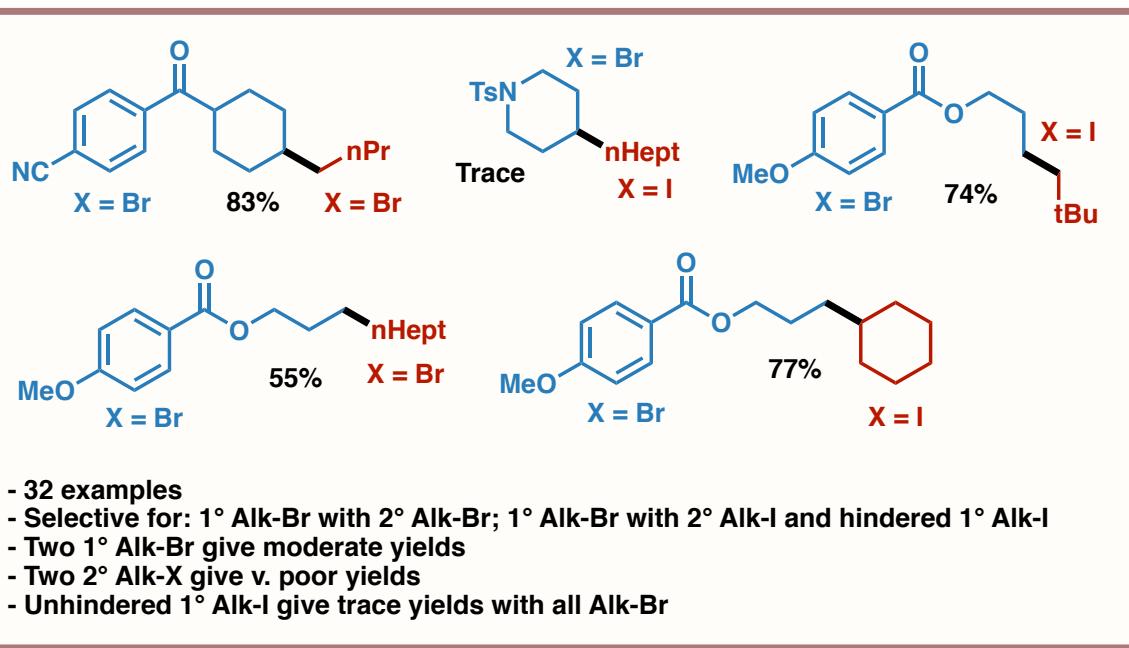
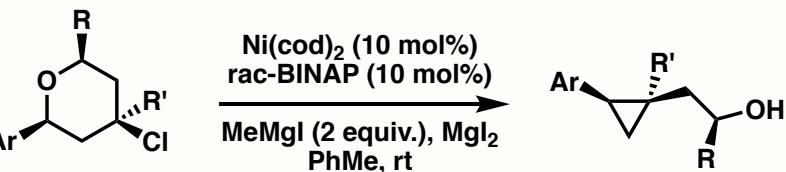
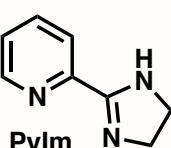
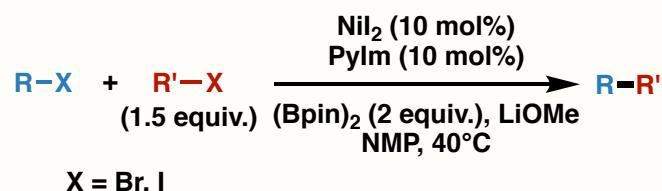
Sequential OA/reduction mechanism intercepted by migratory insertion

Shu. JACS 2019, 7637

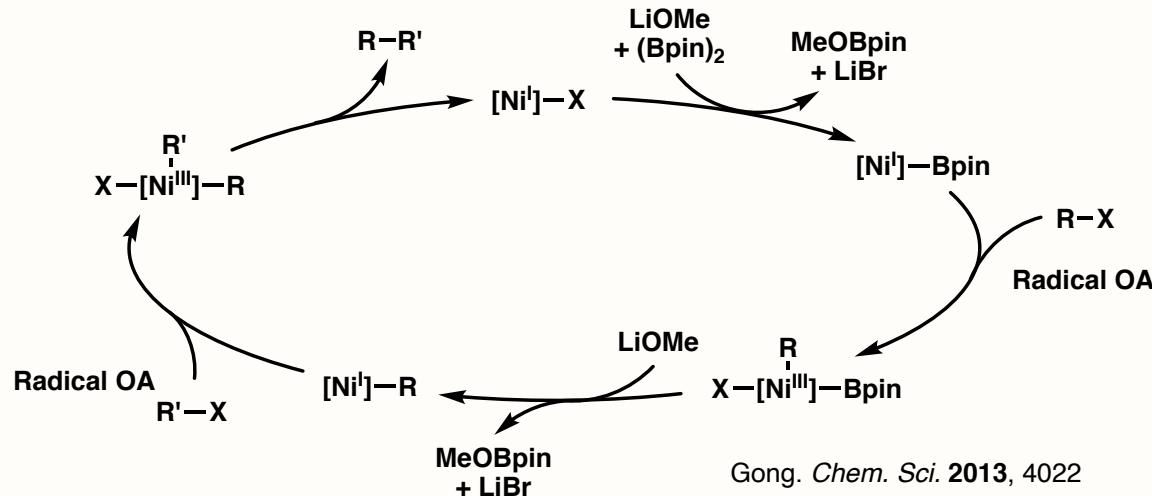


Gong. Org. Lett. 2011, 2138

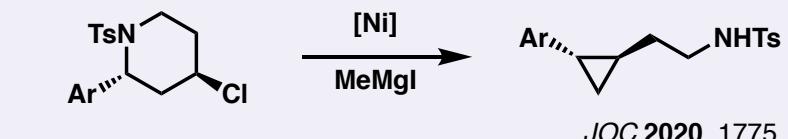
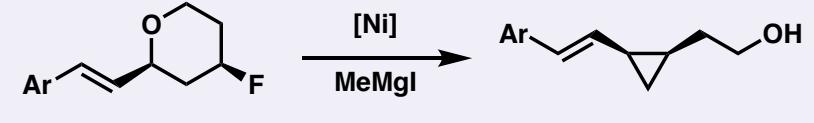
Cross electrophile coupling



Jarvo. JACS 2015, 9760

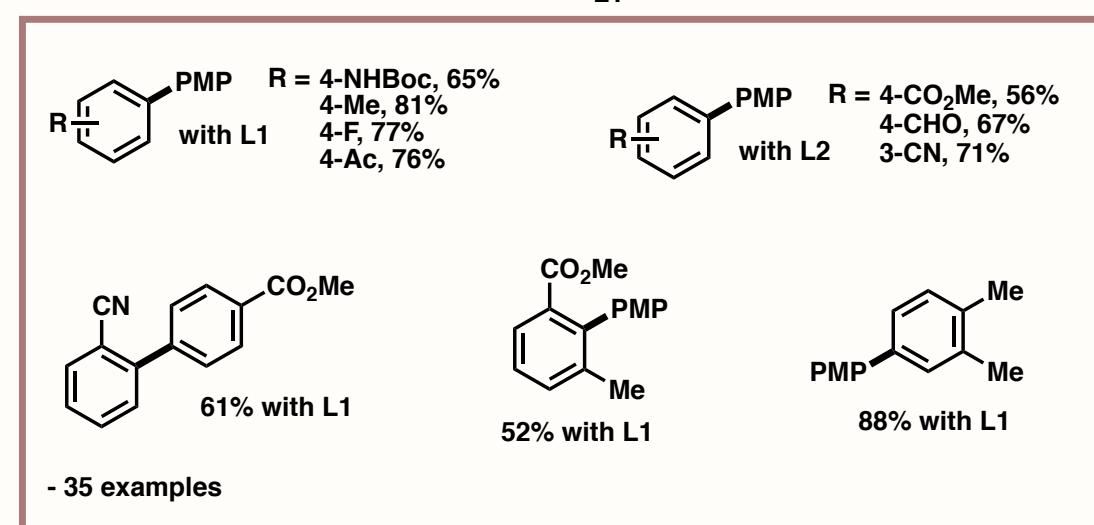
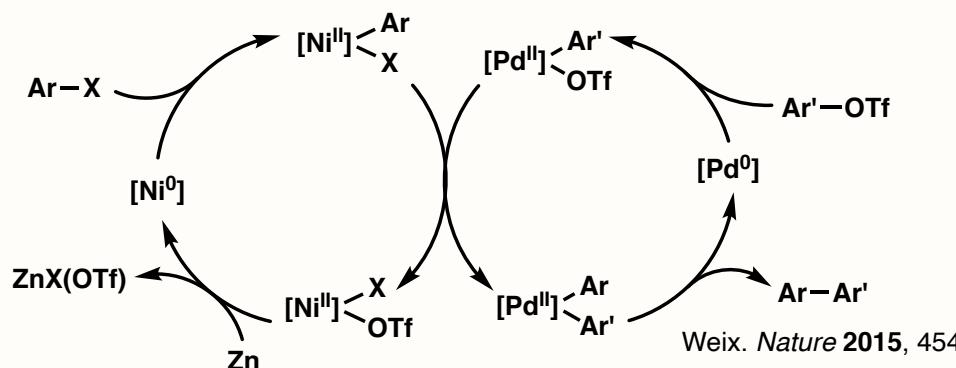
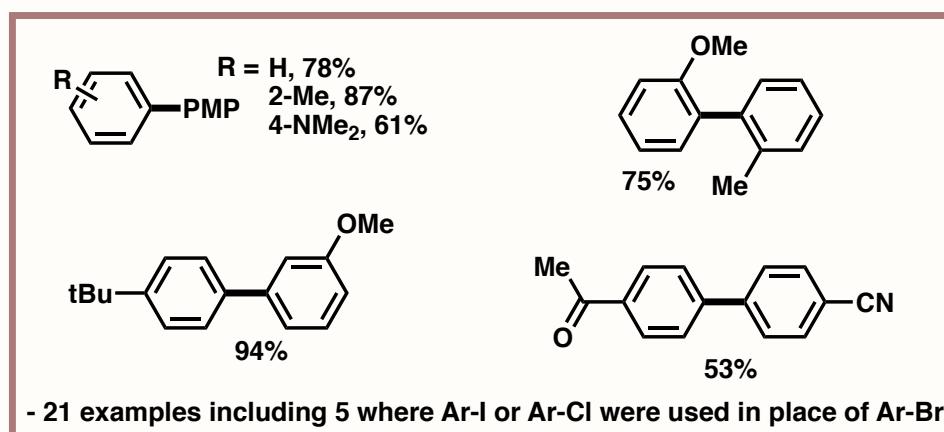
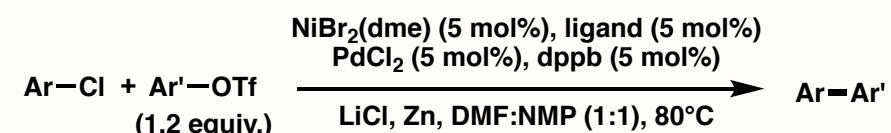
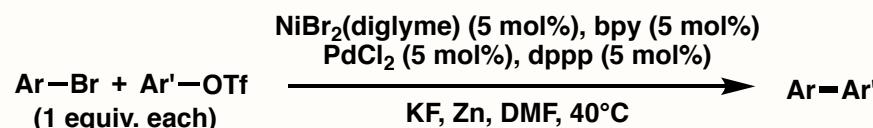
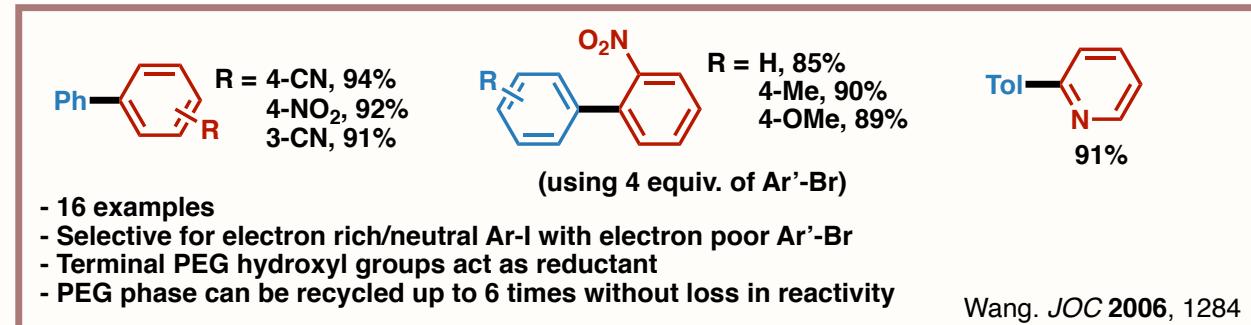
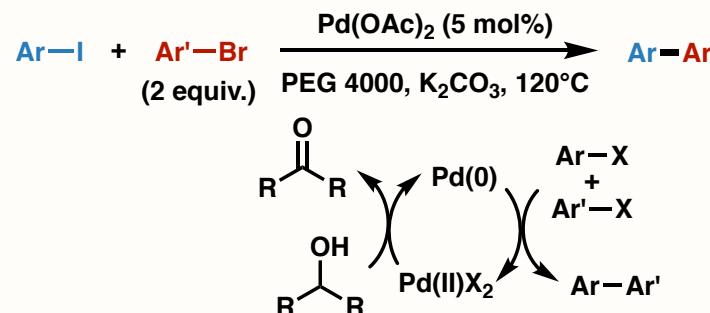


Similar methods by Elizabeth Jarvo

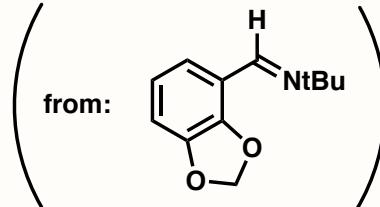
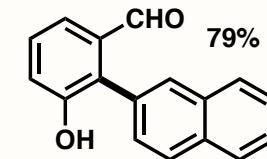
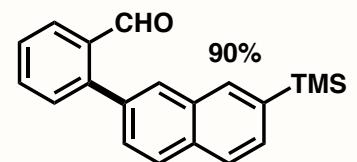
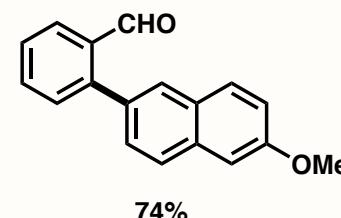
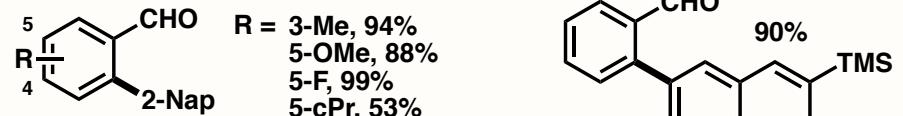
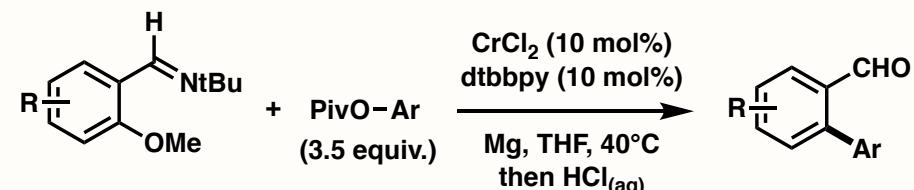
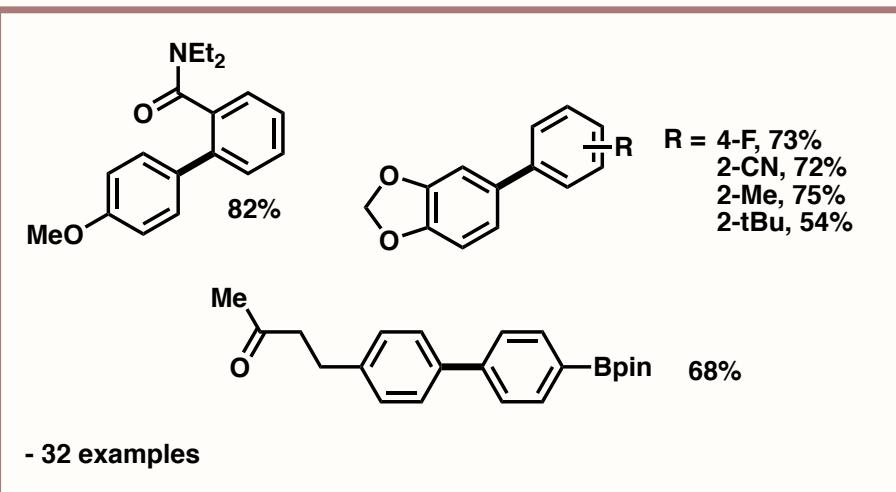
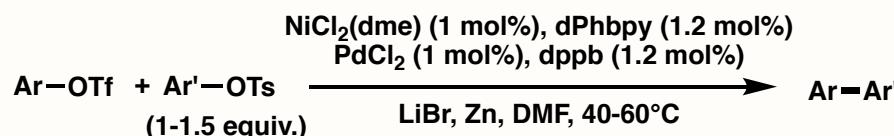


Cross electrophile coupling

Aryl-aryl

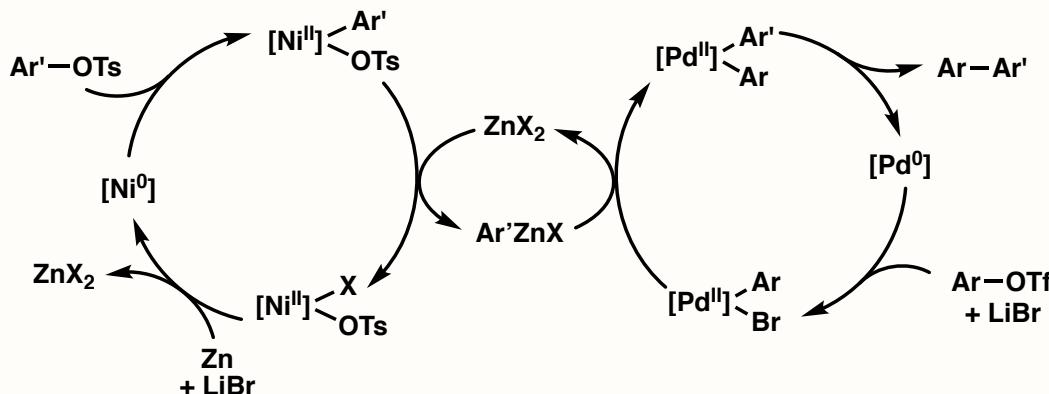


Cross electrophile coupling



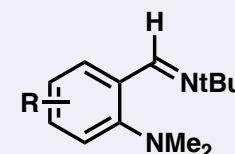
- 67 examples
- OPiv partner appears mostly limited to napthyl and similar polyaromatics
- Proposed mechanism starts from Cr(0) and proceeds by sequential OA/reduction

Zeng. JACS 2020, 7715



- XEC proceeds in diminished yield with Ni only (58% vs 76%)
- Inverted OA selectivity for Ni cat.
- Pd helps to consume Ar'ZnX and improves cross selectivity

Weix. JACS 2020, 10634

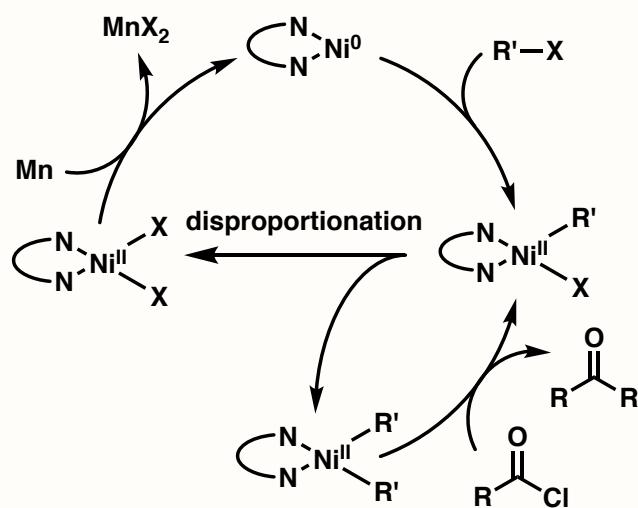
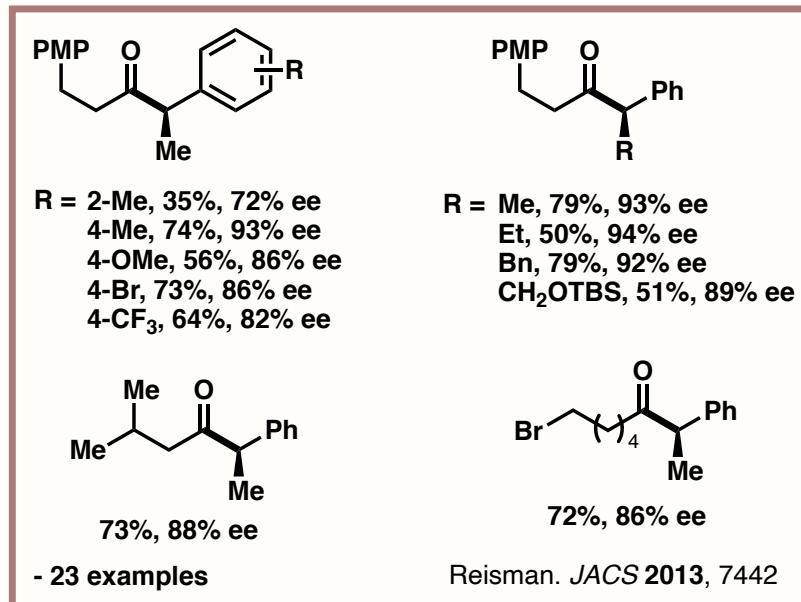
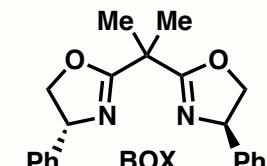
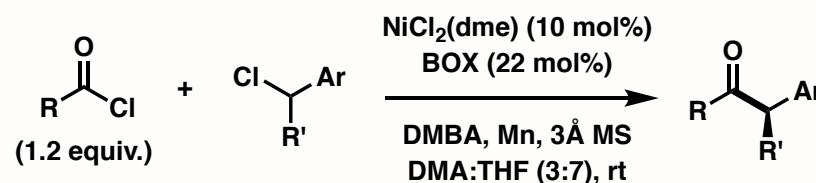
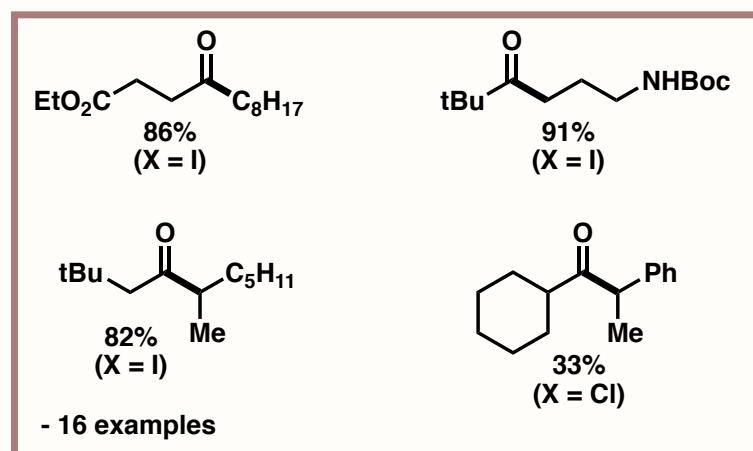
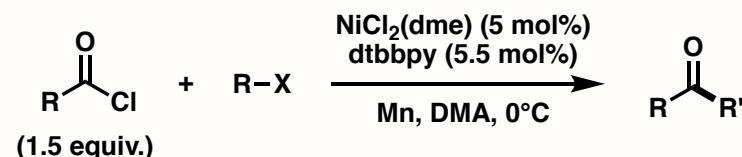


- Follow up report couples dimethyl anilines (C-N bond activation) with pivalates
- Identical conditions
- Similar substrate scope

JACS 2020, 12834

Cross electrophile coupling

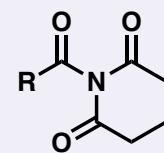
Acylation



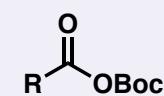
Weix. *Org. Lett.* 2012, 1476

Sequential OA/reduction mechanism starting with OA of acid chloride

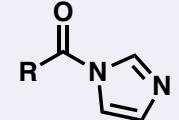
Other activated acyl groups used in XEC



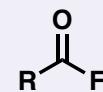
Org. Lett. 2017, 2536



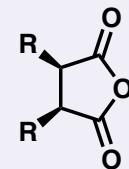
Chem. Commun. 2012, 7034



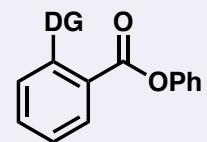
ACS Catal. 2020, 3895



Angew. Chem. Int. Ed. 2020, 13484

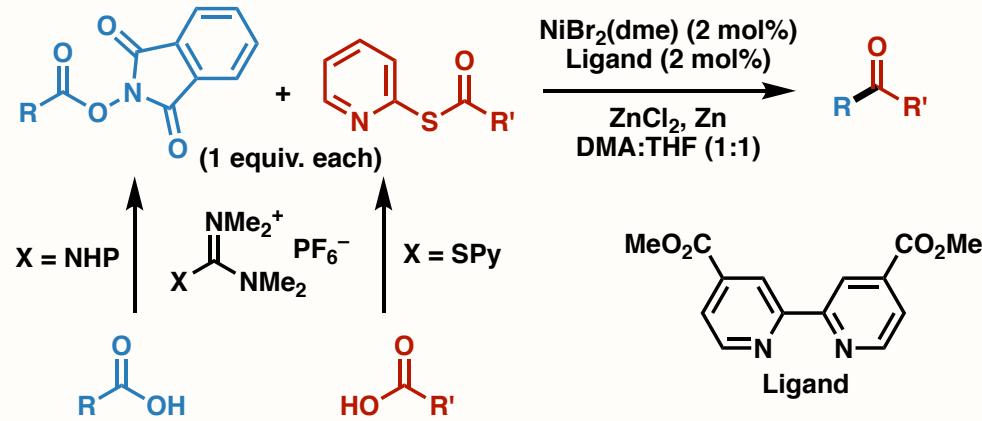


Org. Lett. 2018, 1191

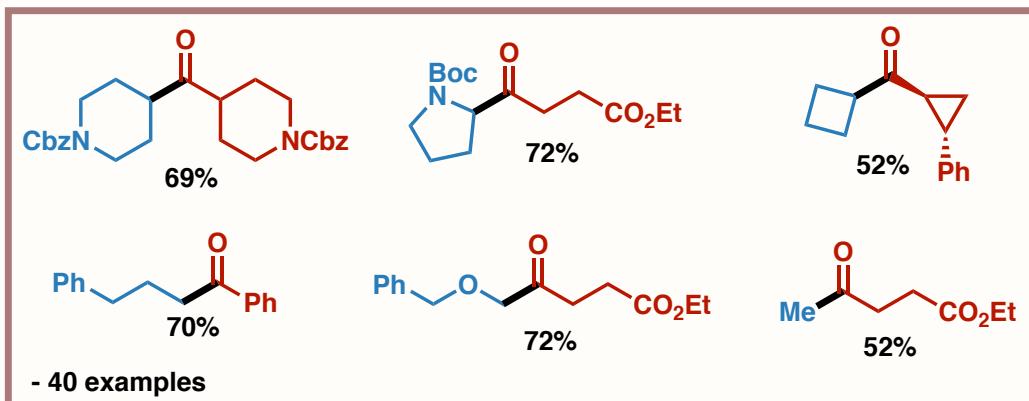


Org. Lett. 2020, 9203

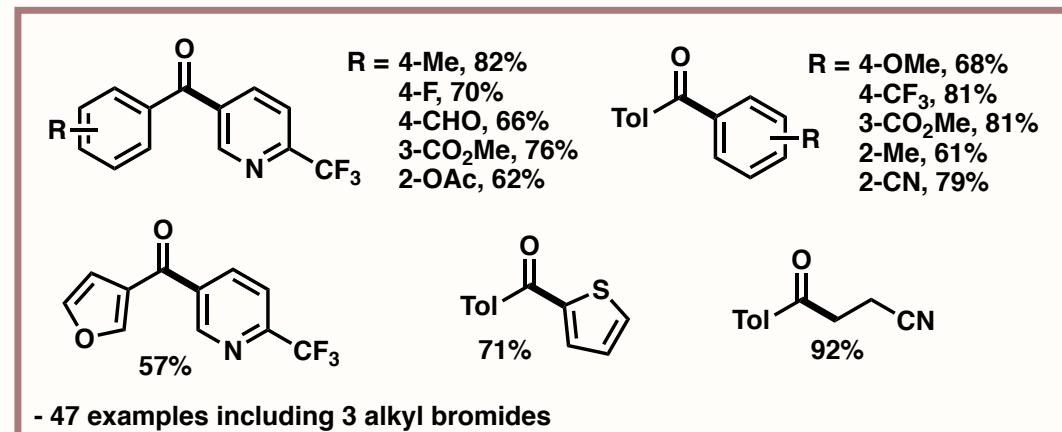
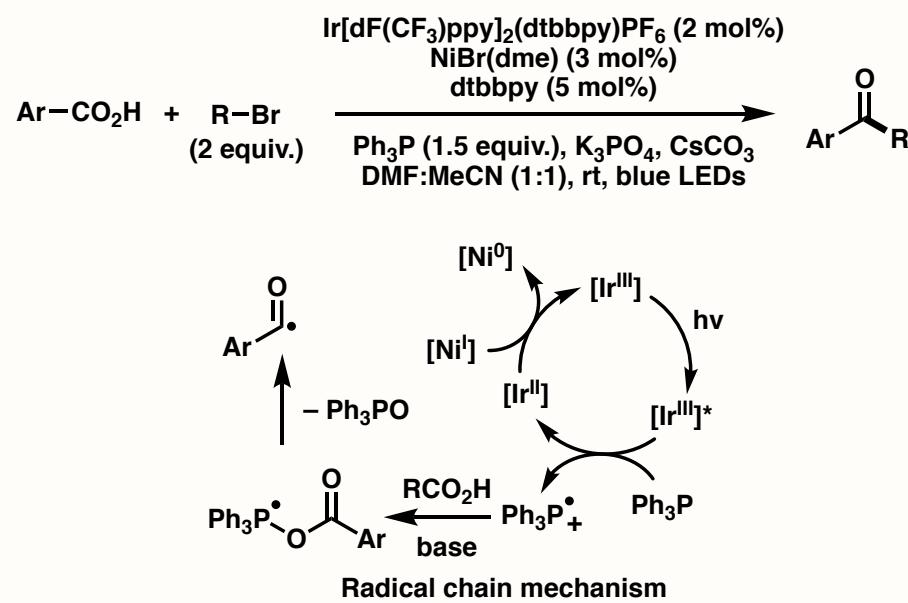
Cross electrophile coupling



Filtration (SiO_2) and concentration gives minimal yield loss in XEC vs. purified reagents



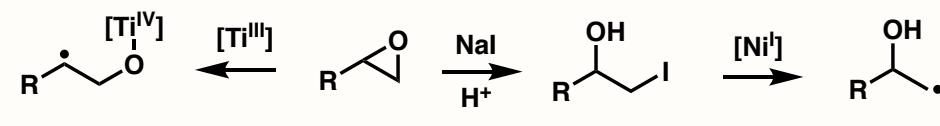
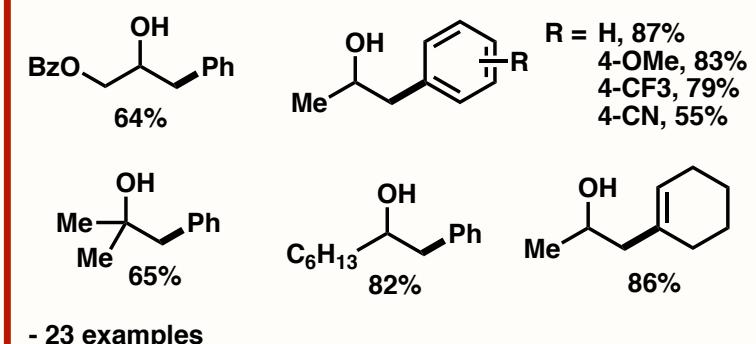
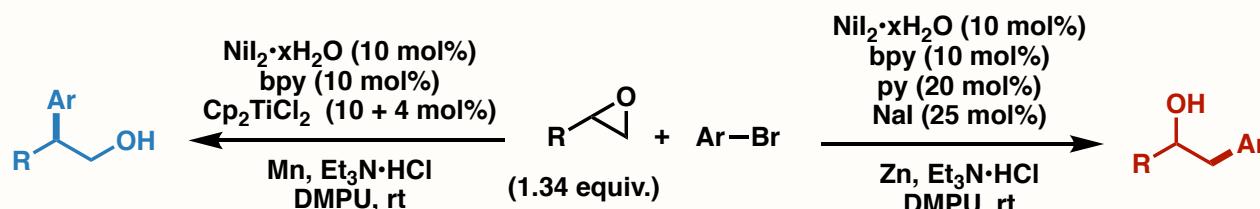
Weix. *Angew. Chem. Int. Ed.* 2019, 12081



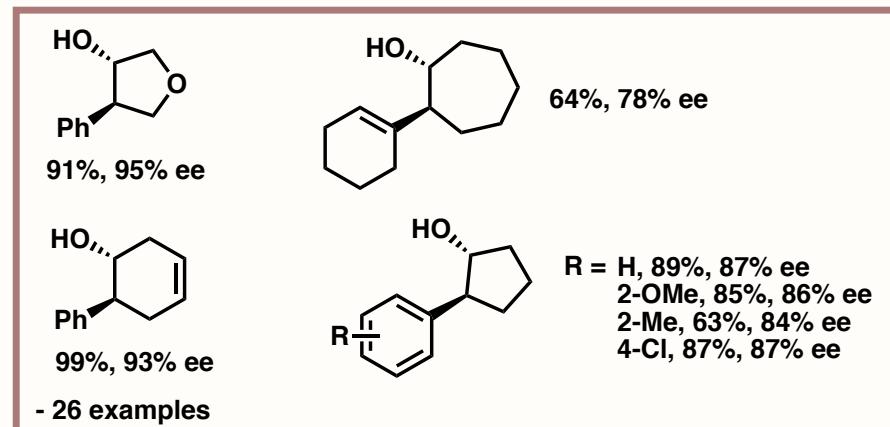
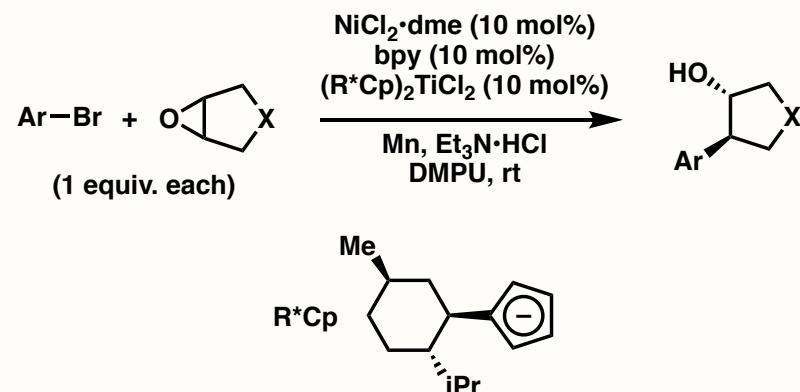
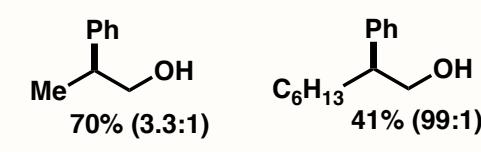
Xie. *Nat. Commun.* 2020, 3312

Cross electrophile coupling

Epoxide/aziridine opening

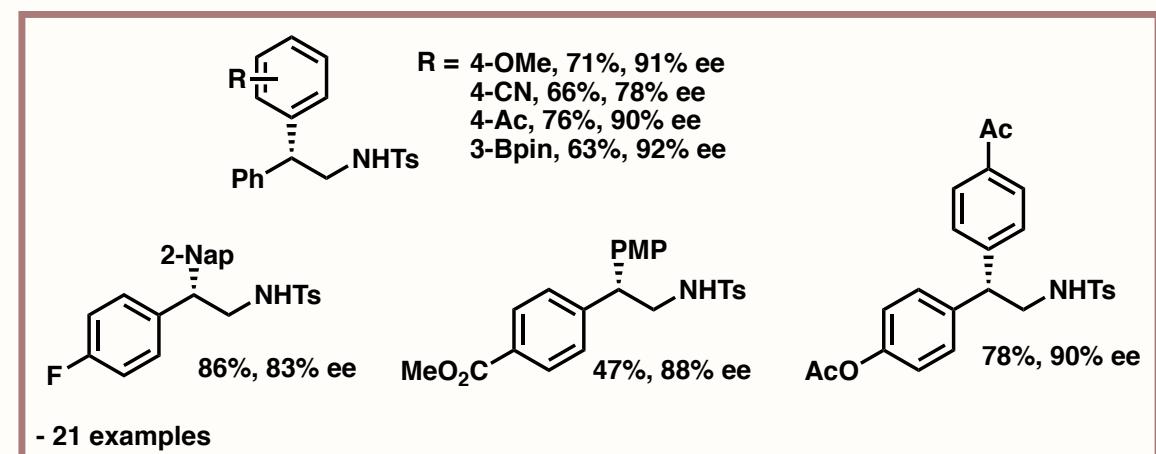
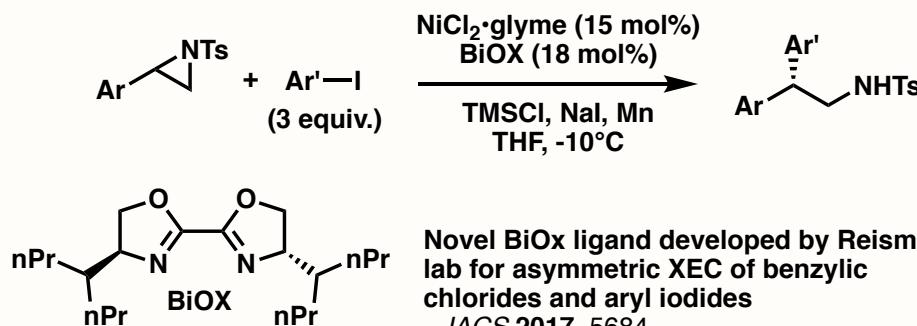


Weix. JACS 2014, 48



Weix. JACS 2015, 3237

Cross electrophile coupling



Simpler, racemic version run at rt also disclosed

Doyle. JACS 2017, 5688

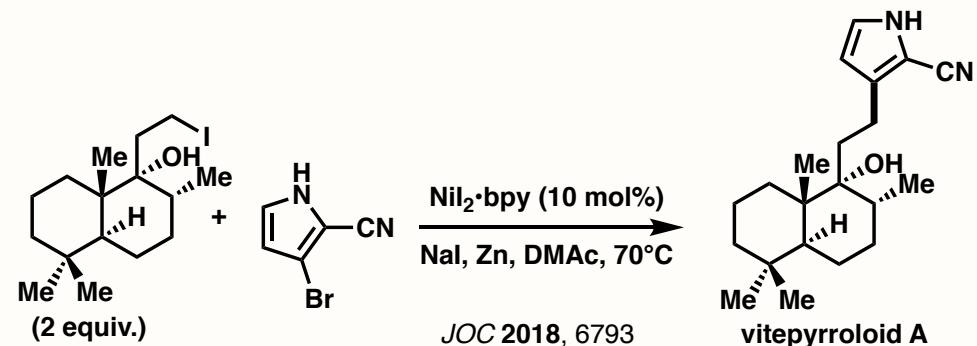
Conclusions and outlook

XEC is field that has exploded in the last decade

Holds many advantages of traditional cross coupling, mainly ubiquity of substrates and orthogonality of scope

Main challenge is selectivity

- xs reagent can be employed
- stereoelectronic matching
- selective mechanistic pathways: radical chain and sequential OA/reduction



Future work

- engage more chloride substrates
- make more hindered bonds (3° with 2°; 3° with 3°)
- lower equivalency of substrate (approaching 1:1)
- different metal catalysts
- use in total synthesis (currently limited)
- more intramolecular couplings/cyclizations

